Diastereoselective Synthesis of Dispirobarbiturates through Et₃N-Catalyzed [3 + 2] Cycloaddition of Barbiturate-Based Olefins with 3-Isothiocyanato Oxindoles

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Supporting Information

ABSTRACT: Under catalysis of 10 mol % of Et_3N , the [3 + 2] cycloaddition of barbiturate-based olefins with 3-isothiocyanato oxindoles underwent smoothly and afforded the desired dispirobarbiturates in up to 99% yield with up to 99:1 dr. The relative configuration of the dispirobarbiturates was unambiguously determined by X-ray single-crystal structure analysis. The reaction mechanism was proposed to shed light on the disatereoselective formation of the dispirobarbiturates.

S pirobarbiturates constitute a class of biologically and medicinally important chemical entities, which have a wide range of biological activities, as shown in Figure 1.¹ Since



Figure 1. Representative bioactive spirobarbiturates.

the potential biological properties of the spirobarbiturates, many efforts were involved in the synthesis of a variety of structurally diverse spirobarbiturates.² For instances, Yoder and co-workers reported the first synthesis of cyclohexanespirobarbiturates in 1921.³ Since then, a number of synthetic methodologies have been developed for the synthesis of spirobarbiturates bearing various cyclic skeletons.⁴ It was worthy to note that only a limited number of examples have been reported on the synthesis of dispirobarbiturates.⁵ Moreover, the stereoselective construction of dispirobarbiturates has not been found in the literature works so far. Therefore, it is highly demanded to develop powerful and efficient protocols for the stereoselective construction of structurally and stereochemically diverse dispirobarbiturates.



By means of [3 + 2] cycloaddition, the Girgis⁶ and Soleimani⁷ groups accomplished the synthesis of pyrazolespirobarbiturates and isoxazole-spirobarbiturates starting from barbiturate-based olefins, as depicted in Scheme 1, respectively. Moreover, pioneered by the Yuan group,^{8a} 3-isothiocyanato oxindoles have found many applications in the diastereoselective and enantioselective synthesis of structurally and

Scheme 1. Synthesis of Spirobarbiturates via [3 + 2] Cycloadditions

Previous works6-7



Received: August 5, 2015 Published: September 24, 2015

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The Journal of Organic Chemistry

stereochemically diverse spirooxindoles.⁸ In this work, we report the first diastereoselective synthesis of oxindolepyrrolidine-dispirobarbiturates, as shown in Scheme 1. Under catalysis of 10 mol % of Et₃N, the [3 + 2] cycloaddition of barbiturate-based olefins with 3-isothiocyanato oxindoles proceeded readily, thus giving the desired dispirobarbiturates in excellent yields and diastereoselectivities. So far, there has been no such work in the literature.

As outlined in Table 1, at the outset, we investigated the effect of basic additives on the chemical yield and



Me O O N Me O 1a	^{Ph} +	$NCS = \frac{\text{additive}}{(10 \text{ mol})}$	$\stackrel{\text{Me}}{} \stackrel{\text{Me}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{} \stackrel{\text{N}}{ \stackrel{\text{N}}{ \text{$	NH S 3aa
entry	additive	time (min)	yield (%) ^b	dr ^c
1	Et ₃ N	10	98	97:3
2	Na ₂ CO ₃	1	84	97:3
3	DBU	13	98	97:3
4	DABCO	8	78	99:1
5	DIPEA	15	91	94:6
6	Na ₂ CO ₃	10	40	97:3
7	DBU	10	75	97:3
8	DABCO	10	60	99:1
9	DIPEA	10	83	94:6

^{*a*}Reactions were carried out with 0.1 mmol of 1a and 0.1 mmol of 2a in the presence of 10 mol % of additive in 1.0 mL of CH₂Cl₂ at room temperature. ^{*b*}Isolated yield. ^{*c*}Determined by ¹H NMR spectroscopy.

diastereoselectivity of the [3 + 2] cycloaddition of 1a and 2a (entries 1-5). In all the cases, the [3 + 2] cycloaddition provided 3aa with excellent diastereoselectivities. With respect to chemical yield, it was affected significantly by the basic additive used. For instance, using DABCO as an additive delivered 3aa in 78% yield. Substitution of DABCO with Na₂CO₂ increased the chemical yield from 78% to 84% (entries 2 vs 4). In the cases of Et₃N, DBU, and DIPEA, all of them gave 3aa in >90% chemical vield (entries 1, 3, and 5). Furthermore, we studied the effect of the bases on the [3 + 2] cycloaddition in the same reaction time (entries 1 and 6-9). As for Na₂CO₃ and DBU, they gave 3aa in the same diastereoselectivities; however, the chemical yield of 3aa differed drastically (entries 6 and 7). In regard to DABCO, it provided 3aa in 60% yield with 99:1 dr (entry 8). In contrast with the former cases, the use of DIPEA increased the chemical yield and lowered the diastereoselectivity (entries 9 vs 6-8). Given the reaction rate, chemical yield, and diastereoselectivity, we chose Et₃N as the optimal base in the [3 + 2] cycloaddition (entry 1).

Next, in the presence of 10 mol % of Et_3N , we screened the solvent effect on the chemical yield and diastereoselectivity of the [3 + 2] cycloaddition of **1a** and **2a**, as summarized in Table 2 (entries 1–6). In all the cases, product **3aa** was obtained in excellent diastereoselectivities. In contrast, the chemical yield changed dramatically with the organic solvents examined. In the case of THF, the [3 + 2] cycloaddition gave **3aa** in 16% yield (entry 4). In comparison with THF, toluene produced a better yield (entries 2 vs 4). The significant increase in the chemical yield was observed with other solvents examined in contrast with the previous two cases (entries 1, 3, and 5 vs 2 and 4).



Me O O N Me O 1a	^{Ph} +	$ \begin{array}{c} \text{NCS} & \text{Et}_3\text{N} \\ Image of the set of th$	I%) r.t. O≓ N Me O	NH S 3aa
entry	solvent	time (min)	yield (%) ^b	dr ^c
1	CHCl ₃	25	>99	97:3
2	toluene	10	47	99:1
3	MeOH	2	88	97:3
4	THF	30	16	98:2
5	MTBE ^d	2	88	97:3
6	CH_2Cl_2	10	98	97:3
7	CHCl ₃	10	65	97:3
8	MeOH	10	88	97:3
9	THF	10	95	98:2
10	$MTBE^{d}$	10	88	97:3

^{*a*}Reactions were carried out with 0.1 mmol of 1a, and 0.1 mmol of 2a in the presence of 10 mol % of Et₃N in 1.0 mL of solvent at room temperature. ^{*b*}Isolated yield. ^{*c*}Determined by ¹H NMR spectroscopy. ^{*d*}Methyl *tert*-butyl ether.

Moreover, in the same reaction time, we examined the effect of the organic solvents on the chemical yield and diastereoselectivity of the [3 + 2] cycloaddition (entries 2 and 6–10). With respect to toluene, CHCl₃, MeOH, and MTBE as solvents, they furnished **3aa** in 47–88% yields with excellent diastereoselectivities (entries 2, 7–8, and 10). Satisfyingly, we found that using both CH₂Cl₂ and THF as solvent delivered **3aa** in excellent chemical yields and diastereoselectivities (entries 6 and 9). Even though THF could provide the excellent chemical yield and diastereoselectivity, it lowered the chemical yield of **3aa** dramatically as the reaction time prolonged because of the formation of some unidentified side products (entries 4 vs 9). After considering the reaction rate, chemical yield, and diastereoselectivity, we determined CH₂Cl₂ as the optimal reaction medium (entry 6).

At last, under the optimal reaction conditions (1:1 molar ratio of 1 to 2, 10 mol % of Et₃N, CH₂Cl₂, r.t.), we extended the reaction scope of the [3 + 2] cycloaddition by using structurally diverse barbiturate-based olefins 1 and 3isothiocyanato oxindoles 2, as presented in Table 3. In most cases, the [3 + 2] cycloaddition processed smoothly and yielded oxindole-pyrrolidine-spirobarbiturates 3 in excellent chemical yields and diastereoselectivities (entries 1-2, 4-5, 8-14, 16-17, and 20). As for the other cases, the chemical yields ranged from 53% to 89%, and diastereoselectivities changed from 91:9 to 98:2 (entries 3, 6-7, 15, and 18-19). The relative configuration of 3fa was determined by the single-crystal Xray analysis (see details in the Supporting Information).⁹ On the basis of the determined relative stereochemistry of 3fa, the relative configuration of the other oxindole-pyrrolidinespirobarbiturates 3 was also assigned similarly, as shown in Table 3. Moreover, with a purpose to gain potentially biologically active new homodimers,¹⁰ we designed and synthesized disulfide-linked homodimer 4 in 60% yield by treating 3aa with DDQ in CH₂Cl₂ at 0 °C, as shown in Scheme 2. Simultaneously, the relative configuration of 4 was also characterized by its single-crystal X-ray analysis (see details in the Supporting Information).¹¹

To interpret the diastereoselective formation of 3fa, the reaction mechanism of the [3 + 2] cycloaddition of 1f and 2a

Table 3. Extension of the Reaction Scope of the [3 + 2] Cycloaddition^{*a*}

($ \begin{array}{cccccccccccccccccccccccccccccccccccc$		$\frac{\text{Et}_{3}\text{N}}{(10 \text{ mol}\%)}$ CH ₂ Cl ₂ r.t.			2
Entry	1 (R ₁)	2 (R ₂ , R ₃)	3	Time (min)	Yield (%) ^b	dr ^c
1	1a (C ₆ H ₅)	2a (H, Me)	3aa	10	98	97:3
2	1a (C ₆ H ₅)	2b (Me, Me)	3ab	1	91	99:1
3	1b (4-MeC ₆ H ₄)	2b (Me, Me)	3bb	1	53	95:5
4	1c(4-FC ₆ H ₄)	2b (Me, Me)	3cb	1	90	94:6
5	1a (C ₆ H ₅)	2c (F, Me)	3ac	1	94	90:10
6	1a (C ₆ H ₅)	2d (H, Bn)	3ad	1	83	96:4
7	1d (2-Thienyl)	2d (H, Bn)	3dd	3	82	98:2
8	1d (2-Thienyl)	2a (H, Me)	3da	3	97	98:2
9	1e (4-BrC ₆ H ₄)	2a (H, Me)	3ea	2	99	97:3
10	1f (4-MeOC ₆ H ₄)	2a (H, Me)	3fa	2	99	96:4
11	1g (4-ClC ₆ H ₄)	2a (H, Me)	3ga	2	99	98:2
12	1h	2a (H, Me)	3ha	2	98	98:2
13	1i (4-CF ₃ C ₆ H ₄)	2a (H, Me)	3ia	2	99	99:1
14	1j (4-FC ₆ H ₄)	2a (H, Me)	3ja	2	99	96:4
15	1k (2-BrC ₆ H ₄)	2a (H, Me)	3ka	5	89	91:9
16	11(3,4,5- <i>tri</i> -MeOC ₆ H ₂)	2a (H, Me)	3la	5	93	97:3
17	1m (3-ClC ₆ H ₄)	2a (H, Me)	3ma	3	94	97:3
18	1n (3-NO ₂ C ₆ H ₄)	2a (H, Me)	3na	3	86	97:3
19	10 (3,4- <i>di</i> -MeOC ₆ H ₃)	2a (H, Me)	30a	1	83	98:2
20	1p (4-NO ₂ C ₆ H ₄)	2a (H, Me)	3pa	3	99	94:6

^{*a*}Reactions were carried out with 0.1 mmol of 1 and 0.1 mmol of 2 in the presence of 10 mol % of Et_3N in 1.0 mL of CH_2Cl_2 at room temperature. ^{*b*}Isolated yield. ^{*c*}Determined by ¹H NMR spectroscopy.

Scheme 2. Dimerization of 3aa via Oxidation Reaction



was predicted as presented in Scheme 3. Deprotonation of 2a with Et_3N led to the formation of enolate 5. Then, the resultant 5 underwent $\begin{bmatrix} 3 + 2 \end{bmatrix}$ cycloaddition with 1f through the transition state TS, thus furnishing the intermediate 6. Finally, the protonation of 6 delivered the diastereoselective formation of 3fa.

In conclusion, we first developed the novel [3 + 2] cycloaddition of barbiturate-based olefins with 3-isothiocyanato oxindoles. The [3 + 2] cycloaddition proceeded readily, thus producing the desired oxindole-pyrrolidine-dispirobarbiturates in excellent chemical yields and diastereoselectivities.

Scheme 3. Proposed Mechanism for the [3 + 2] Cycloaddition



EXPERIMENTAL SECTION

General Information. Unless noted otherwise, all reagents were commercially available and used without further purification. All solvents were distilled from the appropriate drying agents immediately before use. Reactions were monitored by TLC carried out on 0.25 mm SDS silica gel coated glass plates (60F254), and compounds were detected with UV light. The melting point of compounds was determined by a melting point instrument. NMR spectra were recorded on a 400 MHz instrument and calibrated using tetramethylsilane (TMS) as internal reference. IR spectra were monitored with an FT-IR spectrometer using KBr pellets (4000–400 cm⁻¹), and only major peaks were reported in cm⁻¹. High-resolution mass spectra (HRMS) were recorded under electrospray ionization (ESI) conditions on an Orbitrap mass analyzer.

Typical Procedure for the Diastereoselective Synthesis of Dispirobarbiturates 3. Triethylamine (0.01 mmol) was added to a mixture of barbiturate-based olefins 1 (0.1 mmol), and 3-isothiocyanato oxindoles 2 (0.1 mmol) in anhydrous CH_2Cl_2 (1.0 mL). The reaction was stirred at room temperature for 1–10 min. The reaction was quenched with 2.0 mL of saturated aqueous NH_4Cl solution, followed by diluting with 2.0 mL of CH_2Cl_2 . The organic phase was separated, and the aqueous phase was extracted with CH_2Cl_2 (3 × 4 mL). The combined organic phases were dried with anhydrous Na_2SO_4 and concentrated under reduced pressure. The crude product was purified by flash column chromatography on silica gel (petroleum ether/ethyl acetate = 3:1–2:1) to afford the pure products 3 as a white powder (53%–99% yield; 90:10–99:1 dr).

3*aa*. White powder; yield 98% (43.9 mg); mp = 210–211 °C; dr 97:3; ¹H NMR (400 MHz, DMSO): δ 11.54 (s, 1H), 8.26 (d, *J* = 7.6 Hz, 1H), 7.43 (t, *J* = 7.6 Hz, 1H), 7.18 (t, *J* = 7.6 Hz, 1H), 7.12 (d, *J* = 6.8 Hz, 1H), 7.09–7.06 (m, 3H), 6.76 (d, *J* = 7.6 Hz, 2H), 5.39 (s, 1H), 3.27 (s, 3H), 3.17 (s, 3H), 3.14 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.1, 174.5, 166.6, 165.4, 150.4, 144.2, 131.8, 131.3, 129.0, 129.0, 128.6, 128.5, 125.8, 123.4, 109.9, 75.0, 74.3, 57.7, 29.9, 29.4, 27.4; HRMS (ESI) calculated for C₂₃H₂₀N₄O₄S (M + H⁺): 449.12780, found 449.12741; IR (KBr) 3342, 2961, 1730, 1681, 1605, 1491, 1371, 1103, 745 cm⁻¹.

3ab. White powder; yield 91% (42.0 mg); mp = 210 °C; dr 99:1; ¹H NMR (400 MHz, DMSO): δ 11.53 (s, 1H), 8.08 (s, 1H), 7.24 (d, *J* = 7.6 Hz, 1H), 7.14–7.07 (m, 3H), 6.96 (d, *J* = 8 Hz, 1H), 6.76 (d, *J* = 7.6 Hz, 2H), 5.42 (s, 1H), 3.28 (s, 3H), 3.15 (s, 6H), 2.32 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 194.9, 174.4, 166.6, 165.3, 150.5, 141.9, 132.3, 132.0, 131.4, 129.2, 129.0, 128.8, 128.4, 125.9, 109.6, 75.0, 74.4, 67.5, 57.3, 30.0, 29.4, 27.4, 25.6, 21.5; HRMS (ESI) calculated for C₂₄H₂₂N₄O₄S (M + H⁺): 463.14345, found 463.14218; IR (KBr) 3327, 2961, 1730, 1676, 1616, 1501, 1360, 1125, 750 cm⁻¹. **3bb.** White powder; yield 53% (25.2 mg); mp = 194–194.7 °C; dr 95:5; ¹H NMR (400 MHz, DMSO): δ 11.49 (s, 1H), 8.11 (s, 1H), 7.23 (d, *J* = 7.6 Hz, 1H), 6.94 (d, *J* = 8.0 Hz, 1H), 6.88 (d, *J* = 8.0 Hz, 2H), 6.67 (d, *J* = 8.0 Hz, 2H), 5.33 (s, 1H), 3.26 (s, 3H), 3.15 (s, 3H), 3.13 (s, 3H), 2.34 (s, 3H), 2.12 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.2, 174.5,166.7, 165.5, 150.4, 141.9, 137.9, 132.2, 131.3, 129.5, 129.2, 129.0, 128.6, 126.0, 109.5, 75.1, 74.6, 57.7, 29.9, 29.4, 27.4, 21.5, 20.9; HRMS (ESI) calculated for C₂₅H₂₄N₄O₄S (M + H⁺): 477.15910, found 477.15814; IR (KBr) 3305, 2918, 1725, 1670, 1621, 1496, 1355, 1120, 755 cm⁻¹.

3cb. White powder; yield 90% (43.2 mg); mp = 209–209.6 °C; dr 94:6; ¹H NMR (400 MHz, DMSO): δ 11.53 (s, 1H), 8.08 (s, 1H), 7.24 (d, *J* = 7.2 Hz, 1H), 6.96–6.86 (m, 5H), 5.42 (s, 1H), 3.27 (s, 3H), 3.17 (s, 3H), 3.13 (s, 3H), 2.34 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 194.9, 174.4, 166.4, 165.2, 150.5, 141.9, 132.3, 131.5, 131.4, 129.1, 128.0, 125.7, 115.9, 115.7, 109.6, 75.1, 74.4, 56.5, 30.0, 29.5, 27.4, 21.5; HRMS (ESI) calculated for C₂₄H₂₁FN₄O₄S (M + H⁺): 481.13403, found 481.13293; IR (KBr) 3327, 2934, 1735, 1665, 1605, 1496, 1360, 1125, 750 cm⁻¹.

3ac. White powder; yield 94% (43.8 mg); mp = 210–210.3 °C; dr 90:10; ¹H NMR (400 MHz, DMSO): δ 11.59 (s, 1H), 8.12–8.09 (m, 1H), 7.34–7.29 (m, 1H), 7.14–7.11 (m, 4H), 6.73 (d, *J* = 6.4 Hz, 2H), 5.47 (s, 1H), 3.29 (s, 3H), 3.20 (s, 3H), 3.19 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.1, 174.3, 166.3, 165.5, 150.5, 140.6, 131.7, 129.1, 129.0, 128.9, 128.6, 128.4, 117.7, 117.5, 116.3, 116.0, 74.8, 74.2, 56.9, 30.0, 29.5, 27.6; HRMS (ESI) calculated for C₂₃H₁₉FN₄O₄S (M + H⁺): 467.11838, found 467.11758; IR (KBr) 3283, 2967, 1735, 1681, 1616, 1491, 1382, 1103, 755 cm⁻¹.

3ad. White powder; yield 83% (43.5 mg); mp = 213–214 °C; dr 96:4; ¹H NMR (400 MHz, DMSO): δ 11.75 (s, 1H), 8.28 (d, *J* = 7.2 Hz, 1H), 7.33–7.29 (m, 1H), 7.20–7.15 (m, 5H), 7.08–7.03 (m, 4H), 6.86 (d, *J* = 8.0 Hz, 1H), 6.80 (d, *J* = 7.6 Hz, 2H), 5.40 (s, 1H), 5.02 (d, *J* = 16.0 Hz, 1H), 4.83 (d, *J* = 16.0 Hz, 1H), 3.27 (s, 3H), 3.14 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.7, 174.9, 166.6, 165.4, 150.4, 143.0, 135.5, 131.3, 131.2, 129.6, 129.0, 128.9, 128.8, 128.7, 127.9, 127.3, 126.0, 123.4, 110.4, 75.0, 74.7, 58.8, 55.3, 43.7, 29.9, 29.4; HRMS (ESI) calculated for C₂₉H₂₄N₄O₄S (M + H⁺): 525.15910, found 525.15826; IR (KBr) 3348, 3060, 1740, 1676, 1611, 1483, 1367, 1153, 742 cm⁻¹.

3dd. White powder; yield 82% (43.5 mg); mp = 201–201.9 °C; dr 98:2; ¹H NMR (400 MHz, DMSO): δ 11.79 (s, 1H), 8.31 (d, *J* = 7.6 Hz, 1H), 7.35–7.31 (m, 2H), 7.21–7.18 (m, 4H), 7.04–7.02 (m, 2H), 6.84–6.80 (m, 2H), 6.69 (d, *J* = 3.2 Hz, 1H), 5.73 (s, 1H), 5.00(d, *J* = 16.4 Hz, 1H), 4.83 (d, *J* = 16.4 Hz, 1H), 3.28 (s, 3H), 3.15 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.6, 174.8, 166.3, 165.0, 150.5, 143.3, 135.5, 132.0, 131.5, 129.9, 129.1, 128.9, 128.4, 127.8, 127.2, 126.8, 125.7, 123.3, 110.3, 75.4, 74.7, 55.3, 54.7, 43.6, 30.0, 29.4;

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HRMS (ESI) calculated for $C_{27}H_{22}N_4O_4S_2$ (M + H⁺): 531.11552, found 531.11438; IR (KBr) 3321, 3076, 1735, 1670, 1611, 1496, 1371, 1103, 750 cm⁻¹.

3*da*. White powder; yield 97% (44.0 mg); mp = 205 °C; dr 98:2; ¹H NMR (400 MHz, DMSO): δ 11.58 (s, 1H), 8.32–8.30 (m, 1H), 7.44 (t, *J* = 7.6 Hz, 1H), 7.29 (d, *J* = 5.2 Hz, 1H), 7.21 (t, *J* = 7.6 Hz, 1H), 7.04 (d, *J* = 8.0 Hz, 1H), 6.77 (t, *J* = 4.4 Hz, 1H), 6.64 (d, *J* = 3.2 Hz, 1H), 5.64 (s, 1H), 3.28 (s, 3H), 3.14 (s, 6H); ¹³C NMR (100 MHz, DMSO): δ 195.2, 174.4, 166.3, 165.1, 150.4, 144.4, 132.4, 131.6, 129.3, 129.0, 128.1, 126.9, 125.4, 123.3, 109.7, 75.5, 74.4, 54.2, 31.1, 30.0, 29.4, 27.3; HRMS (ESI) calculated for C₂₁H₁₈N₄O₄S₂ (M + H⁺): 455.08422, found 455.08365; IR (KBr) 3354, 3086, 1730, 1676, 1611, 1491, 1371, 1147, 739 cm⁻¹.

3ea. White powder; yield 99% (52.1 mg); mp = 196–196.4 °C; dr 97:3; ¹H NMR (400 MHz, DMSO): δ 11.57 (s, 1H), 8.22 (d, *J* = 7.6 Hz, 1H), 7.46–7.42 (m, 1H), 7.27 (d, *J* = 8.4 Hz, 1H), 7.18 (t, *J* = 7.6 Hz, 1H), 7.08 (d, *J* = 8.0 Hz, 1H), 6.74 (d, *J* = 8.8 Hz, 2H), 5.43 (s, 1H), 3.27 (s, 3H), 3.18 (s, 3H), 3.17 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 194.7, 174.5, 166.2, 165.1, 150.6, 144.2, 131.8, 131.5, 131.4, 131.3, 128.5, 125.5, 123.4, 121.9, 109.9, 74.9, 74.1, 56.1, 30.0, 29.5, 27.4; HRMS (ESI) calculated for C₂₃H₁₉BrN₄O₄S (M + H⁺): 527.03831, found 527.03790; IR (KBr) 3354, 3174, 1725, 1681, 1611, 1485, 1377, 1103, 750 cm⁻¹.

3fa. White powder; yield 99% (47.3 mg); mp = 192–192.7 °C; dr 96:4; ¹H NMR (400 MHz, DMSO): δ 11.49 (s, 1H), 8.30 (d, *J* = 7.6 Hz, 1H), 7.44–7.40 (m, 1H), 7.21 (t, *J* = 7.6 Hz, 1H), 7.03 (d, *J* = 7.6 Hz, 1H), 6.76 (d, *J* = 8.8 Hz, 2H), 6.63 (d, *J* = 8.8 Hz, 2H), 5.21 (s, 1H), 3.60 (s, 3H), 3.24 (s, 3H), 3.12 (s, 3H), 3.12 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.7, 174.6, 166.8, 165.7, 159.5, 150.3, 144.1, 131.2, 130.8, 128.5, 125.9, 123.3, 122.9, 114.3, 109.8, 75.3, 74.7, 66.8, 58.6, 55.4, 29.8, 29.3, 27.3; HRMS (ESI) calculated for C₂₄H₂₂N₄O₅S (M + H⁺): 479.13837, found 479.13745; IR (KBr) 3179, 2972, 1705, 1703, 1611, 1513, 1371, 1093, 755 cm⁻¹.

3ga. White powder; yield 99% (47.7 mg); mp = 205 °C; dr 98:2; ¹H NMR (400 MHz, DMSO): δ 11.56 (s, 1H), 8.24 (d, J = 7.2 Hz, 1H), 7.46–7.42 (m, 1H), 7.18 (t, J = 7.6 Hz, 1H), 7.13 (d, J = 8.4 Hz, 2H), 7.08 (d, J = 7.6 Hz, 1H), 6.82 (d, J = 8.4 Hz, 2H), 5.45 (s, 1H), 3.28 (s, 3H), 3.18 (s, 3H), 3.17 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 194.8, 174.5, 166.2, 165.1, 150.5, 144.2, 133.3, 131.4, 131.2, 130.8, 128.8, 128.5, 125.6, 123.4, 109.9, 75.0, 74.2, 56.2, 30.0, 29.5, 27.4; HRMS (ESI) calculated for C₂₃H₁₉ClN₄O₄S (M + H⁺): 483.08883, found 483.08813; IR (KBr) 3354, 2923, 1725, 1681, 1611, 1491, 1371, 1088, 750 cm⁻¹.

3ha. White powder; yield 98% (49.6 mg); mp = 199–199.6 °C; dr 98:2; ¹H NMR (400 MHz, DMSO): δ 11.49 (s, 1H), 8.29 (d, *J* = 7.2 Hz, 1H), 7.46–7.42 (m, 1H), 7.22 (t, *J* = 7.6 Hz, 1H), 7.07 (d, *J* = 8.0 Hz, 1H), 6.58–6.56 (m, 1H), 6.31–6.30 (m, 2H), 5.19 (s, 1H), 4.08– 4.07 (m, 4H), 3.25 (s, 3H), 3.14 (s, 3H), 3.12 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.4, 174.5, 166.7, 165.5, 150.4, 144.2, 143.8, 143.3, 131.3, 128.5, 125.9, 124.0, 123.3, 122.8, 118.0, 117.4, 109.9, 75.3, 74.5, 64.4, 64.3, 57.8, 29.9, 29.3, 27.4; HRMS (ESI) calculated for C₂₅H₂₂N₄O₆S (M + H⁺): 507.13328, found 507.13257; IR (KBr) 3288, 2929, 1730, 1670, 1605, 1507, 1377, 1125, 755 cm⁻¹.

3*ia*. White powder; yield 99% (51.1 mg); mp = 189 °C; dr 99:1; ¹H NMR (400 MHz, DMSO): δ 11.62 (s, 1H), 8.19 (d, *J* = 7.6 Hz, 1H), 7.47–7.42 (m, 3H), 7.19–7.11 (m, 2H), 6.98 (d, *J* = 8.4 Hz, 1H), 5.65 (s, 1H), 3.30 (s, 3H), 3.21 (s, 3H), 3.21 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 194.1, 174.4, 165.9, 164.8, 150.7, 144.3, 137.1, 131.5, 129.9, 128.5, 125.6, 125.6, 125.4, 123.5, 110.0, 74.9, 73.8, 55.1, 30.1, 29.5, 27.5; HRMS (ESI) calculated for C₂₄H₁₉F₃N₄O₄S (M + H⁺): 517.11519, found 517.11414; IR (KBr) 3179, 2956, 1733, 1698, 1611, 1469, 1327, 1118, 752 cm⁻¹.

3*ja*. White powder; yield 99% (46.1 mg); mp = 209–209.7 °C; dr 96:4; ¹H NMR (400 MHz, DMSO): δ 11.55 (s, 1H), 8.26 (d, *J* = 7.6 Hz, 1H), 7.46–7.41 (m, 1H), 7.22–7.18 (m, 1H), 7.06 (d, *J* = 7.6 Hz, 1H), 6.94–6.84 (m, 4H), 5.40 (s, 1H), 3.27 (s, 3H), 3.16 (s, 3H), 3.15 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.1, 174.5, 166.4, 165.3, 150.5, 144.2, 131.6, 131.5, 131.4, 128.5, 127.9, 127.8, 125.7, 123.4, 115.9, 115.6, 109.9, 75.1, 74.4, 56.7, 30.0, 29.4, 27.4; HRMS (ESI)

calculated for $C_{23}H_{19}FN_4O_4S~(M+H^{+}):$ 467.11838, found 467.11728; IR (KBr) 3305, 2967, 1730, 1681, 1605, 1474, 1371, 1229, 750 cm $^{-1}.$

3*ka*. White powder; yield 89% (46.8 mg); mp = 203 °C; dr = 91:9; ¹H NMR (400 MHz, DMSO): δ 11.51 (s, 1H), 8.28 (d, *J* = 7.6 Hz, 1H), 7.55 (d, *J* = 8.0 Hz, 1H), 7.40 (t, *J* = 7.6 Hz, 1H), 7.25 (t, *J* = 7.6 Hz, 1H), 7.12–7.08 (m, 1H), 7.02 (d, *J* = 4.4 Hz, 2H), 6.95 (d, *J* = 7.6 Hz, 1H), 5.70 (s, 1H), 3.19 (s, 3H), 3.05 (s, 3H), 2.95 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 196.8, 174.0, 166.7, 166.6, 149.9, 144.0, 133.8, 131.5, 131.5, 131.2, 129.4, 128.4, 127.5, 126.6, 125.3, 123.2, 109.9, 109.7, 75.1, 74.3, 58.6, 29.8, 29.1, 27.3; HRMS (ESI) calculated for C₂₃H₁₉BrN₄O₄S (M + H⁺): 527.03831, found 527.03748; IR (KBr) 3261, 2939, 1735, 1681, 1611, 1469, 1377, 1098, 760 cm⁻¹.

3*a*. White powder; yield 93% (50.0 mg); mp = 206–206.5 °C; dr 97:3; ¹H NMR (400 MHz, DMSO): δ 11.51 (s, 1H), 8.34 (d, *J* = 7.6 Hz, 1H), 7.46 (t, *J* = 7.6 Hz, 1H), 7.25 (t, *J* = 7.6 Hz, 1H), 7.08 (d, *J* = 7.6 Hz, 1H), 6.07 (s, 2H), 5.14 (s, 1H), 3.52 (s, 3H), 3.44 (s, 6H), 3.29 (s, 3H), 3.15 (s, 3H), 3.07 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 195.7, 174.6, 166.9, 165.7, 152.8, 150.3, 144.2, 137.7, 131.3, 128.5, 126.7, 126.2, 123.2, 110.0, 106.7, 75.3, 74.5, 60.4, 59.1, 56.5, 56.1, 29.9, 29.3, 27.4, 19.0; HRMS (ESI) calculated for C₂₆H₂₆N₄O₇S (M + H⁺): 539.15950, found 539.15863; IR (KBr) 3337, 2956, 1735, 1681, 1583, 1463, 1377, 1125, 755 cm⁻¹.

3*ma*. White powder; yield 94% (45.3 mg); mp = 192–193 °C; dr 97:3; ¹H NMR (400 MHz, DMSO): δ 11.59 (s, 1H), 8.24 (d, *J* = 7.6 Hz, 1H), 7.46 (t, *J* = 7.6 Hz, 1H), 7.24–7.19 (m, 2H), 7.15–7.07 (m, 2H), 6.85–6.80 (m, 2H), 5.44 (s, 1H), 3.27 (s, 3H), 3.17 (s, 3H), 3.16 (s, 3H): ¹³C NMR (100 MHz, DMSO): δ 194.7, 174.5, 166.1, 165.0, 150.5, 144.2, 134.1, 133.3, 131.5, 130.7, 129.2, 128.6, 128.5, 125.6, 123.4, 110.0, 75.0, 74.1, 56.1, 30.1, 29.4, 27.4; HRMS (ESI) calculated for C₂₃H₁₉ClN₄O₄S (M + H⁺): 483.08883, found 483.08774; IR (KBr) 3299, 2961, 1741, 1676, 1611, 1469, 1371, 1109, 745 cm⁻¹.

3*na*. White powder; yield 86% (42.4 mg); mp = 167 °C; dr 97:3; ¹H NMR (400 MHz, DMSO): δ 11.66 (s, 1H), 8.23 (d, J = 7.6 Hz, 1H), 7.98 (d, J = 8.0 Hz, 1H), 7.66 (s, 1H), 7.47–7.34 (m, 3H), 7.21 (t, J = 7.6 Hz, 1H), 7.10 (d, J = 7.6 Hz, 1H), 5.66 (s, 1H), 3.29 (s, 3H), 3.19 (s, 3H), 3.18 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 194.4, 174.5, 165.8, 164.8, 150.7, 147.7, 144.2, 136.7, 134.0, 131.6, 130.3, 128.5, 125.3, 124.2, 123.5, 123.3, 110.0, 75.0, 74.0, 55.1, 30.2, 29.5, 27.4; HRMS (ESI) calculated for C₂₃H₁₉N₅O₆S (M + H⁺): 494.11288, found 494.11188; IR (KBr) 3218, 2956, 1710, 1681, 1611, 1529, 1349, 1103, 750 cm⁻¹.

30a. White powder; yield 83% (42.2 mg); mp = 206–207 °C; dr 98:2; ¹H NMR (400 MHz, DMSO): δ 11.51 (s, 1H), 8.35 (d, *J* = 7.6 Hz, 1H), 7.46–7.42 (m, 1H), 7.23 (t, *J* = 7.6 Hz, 1H), 7.04 (d, *J* = 8.0 Hz, 1H), 6.68 (d, *J* = 8.4 Hz, 1H), 6.42–6.40 (m, 1H), 6.34 (s, 1H), 5.15 (s, 1H), 3.61 (s, 3H), 3.33 (s, 3H), 3.25 (s, 3H), 3.12 (s, 3H), 3.10 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 196.0, 174.7, 166.9, 165.8, 150.3, 149.1, 148.4, 144.2, 131.3, 128.5, 126.2, 123.2, 123.0, 122.6, 112.3, 111.7, 109.9, 75.3, 74.7, 59.2, 55.7, 55.4, 29.8, 29.3, 27.3; HRMS (ESI) calculated for C₂₅H₂₄N₄O₆S (M + H⁺): 509.14893, found 509.14767; IR (KBr) 3342, 2956, 1735, 1677, 1611, 1485, 1371, 1147, 755 cm⁻¹.

3pa. White powder; yield 99% (48.8 mg); mp = 198 °C; dr 94:6; ¹H NMR (400 MHz, DMSO): δ 11.66 (s, 1H), 8.16 (d, J = 7.6 Hz, 1H), 7.90 (d, J = 8.8 Hz, 2H), 7.46 (t, J = 7.6 Hz, 1H), 7.18–7.13 (m, 2H), 7.01 (d, J = 8.4 Hz, 2H), 5.71 (s, 1H), 3.31 (s, 3H), 3.23 (s, 3H), 3.22 (s, 3H); ¹³C NMR (100 MHz, DMSO): δ 193.8, 174.3, 165.7, 164.7, 150.7, 147.1, 144.3, 140.0, 131.6, 130.2, 128.4, 125.2, 123.7, 123.6, 110.1, 74.9, 73.7, 54.7, 30.2, 29.6, 27.5; HRMS (ESI) calculated for C₂₃H₁₉N₅O ₆S (M + H⁺): 494.11288, found 494.11169; IR (KBr) 3152, 2961, 1707, 1681, 1611, 1529, 1349, 1120, 750 cm⁻¹.

Typical Procedure for the Synthesis of Compound 4. DDQ (0.1 mmol) was added to the solution of dispirobarbiturate 3aa (0.1 mmol) in CH₂Cl₂ (1.0 mL). The reaction was stirred at 0 °C for 2 h. The reaction was quenched with 2.0 mL of saturated aqueous NaHSO₃ solution, followed by diluting with 2.0 mL of CH₂Cl₂. The organic phase was separated, and the aqueous phase was extracted with CH₂Cl₂ (3 × 4 mL). The combined organic phases were washed with brine (3 × 10 mL) and dried with anhydrous Na₂SO₄, and concentrated under reduced pressure. The resulted residue was

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purified by flash column chromatography on silica gel (petroleum ether/ethyl acetate = 1:1) to afford 4 as a white powder; yield 60% (53.6 mg); mp = 220 °C; ¹H NMR (400 MHz, DMSO): δ 8.27–7.94 (m, 1H), 7.43–7.34 (m, 1H), 7.21–7.11 (m, 4H), 6.96–6.75 (m, 3H), 5.38–4.48 (m, 1H), 3.27–2.84 (m, 9H); ¹³C NMR (100 MHz, DMSO): δ 174.5, 174.3, 174.2, 169.1, 168.8, 167.9, 167.5, 166.4, 165.9, 150.3, 144.0, 130.9, 130.4, 130.4, 130.2, 129.8, 128.9, 128.6, 126.2, 126.1, 123.0, 109.7, 85.7, 85.5, 75.0, 74.3, 74.3, 67.3, 67.2, 29.9, 29.4, 29.3, 29.2, 29.0, 28.9, 27.4; HRMS (ESI) calculated for C₄₆H₃₈N₈O₈S₂ (M + H⁺): 895.23268, found 895.23175; IR (KBr) 2961, 1730, 1682, 1599, 1447, 1371, 1093, 750 cm⁻¹.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b01810.

Crystallographic data for 3fa (CIF)

Crystallographic data for 4 (CIF)

Copies of NMR for dispirobarbiturates 3 and 4; X-ray single-crystal structure analysis data for 3fa and 4 (PDF)

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank the Beijing Municipal Commission of Education (No. JC015001200902), the Beijing Municipal Natural Science Foundation (No. 7102010, No. 2122008), the Basic Research Foundation of Beijing University of Technology (X4015001201101), the Funding Project for Academic Human Resources Development in Institutions of Higher Learning Under the Jurisdiction of Beijing Municipality (No. PHR201008025), and the Doctoral Scientific Research Start-up Foundation of Beijing University of Technology (No. 52015001200701) for financial support.

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(11) CCDC 1416436 contains the supplementary crystallographic data for compound 4. These data can be obtained free of charge from The Cambridge Crystallographic Data Center via www.ccdc.cam.ac. uk/data_request/cif.